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gas law: Unlike the other gas laws we talked about, the ideal gas law doesn't describe what happens to a gas when you manipulate it (i.e. when you change the pressure, volume, temperature). Instead, the ideal gas law describes how a gas will behave under some unchanging set of conditions referred to as an equation of state. The ideal gas law | The Cavalcade o' Chemistry We're now posting original research! Yes, as of late November we are hosting our own

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nRT or $P_1 V_1 = P_2 V_2 T_1 T_2$ Use your knowledge of the ideal and combined gas laws to solve the following problems. If it involves moles or grams, it must be $PV = nRT$ 1) If four moles of a gas at a pressure of 5.4 atmospheres have a volume of 120 liters, what is the temperature? 1973 K Ideal Gas Law Worksheet $PV = nRT$ Posts about Practice worksheets written by misterguch. The Cavalcade o' Chemistry. Celebrating 20 years of chemistry goodness. Seriously,

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 The Ideal and Combined Gas Laws
 $PV = nRT$ or $P_1V_1 = P_2V_2$
 $\frac{P_1V_1}{T_1} = \frac{P_2V_2}{T_2}$
 Use your knowledge of the ideal and combined gas laws to solve the following problems. If it involves moles or grams, it must be
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 temperature is 30 C, what will the volume of the gas inside

be if the hull of the submarine breaks? 4)
 People who are angry sometimes say that they feel as if they'll explode. If a calm person with a lung capacity of 3.5 liters and a body temperature of 360 C gets angry, what will the volume of the person's lungs be if their
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 standard temperature and
 pressure is compressed to
 473 mL.Gas law packet
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 Directions: Examine each
 question and then write

the formula of the gas law
 you plan to use to solve
 each question. Show
 which values you are
 given, which values are
 unknown or which values
 need to be calculated.v,
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 gases ...
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Mixed Gas Laws

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metal tank contains three
gases ...

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MIXED GAS LAWS
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Examine each question
and then write the

formula of the gas law
you plan to use to solve
each question. Show
which values you are
given, which values are
unknown or which values
need to be calculated.

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pressure of 5.4
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volume of 120 liters, what
is the temperature? 1973
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Law to answer the
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standard temperature and
pressure is compressed to
473 mL.

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