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Using periodic table to find electron distributions: Noble Gas Distributions - Find noble gas on end of row before the element and put its symbol in brackets.

Use diagonal rule to write electron distribution of Sn (atomic no. 50): _____

Use noble gas distribution to write electron distribution of Sn: _____ The Chemistry Quiz CR1. CR2. 1. 2 ...

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3-19a,19b-Electron Distributions Wkst-Key

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6. The highly probable location of an electron within the atom is a(n) _____. II. Write the electron configuration for the following: 1. Mg: _____ 2. As: _____ III. In the space below, show the orbital notation for Mg: p s 3 5 2 opposite spins 2n2 orbital

1s22s22p63s2

1s22s22p63s23p64s23d104p3

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electron distribution-forming ions -Atoms become ions when it has a full outer energy level (8 valence electrons) -Atoms that lose (give away) electrons become positive charged ions - Atoms that gain (steal) electrons become negative charged ions

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